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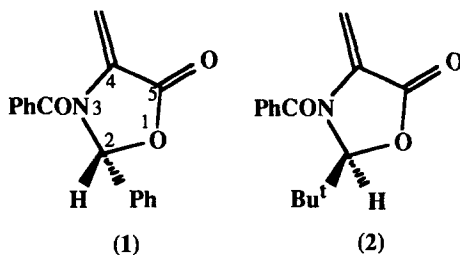
***Exo*-Diastereoselective 1,3-Dipolar Cycloadditions of Azomethine Ylides to (2*R*)-3-Benzoyl-4-methylene-2-phenyloxazolidin-5-one.**

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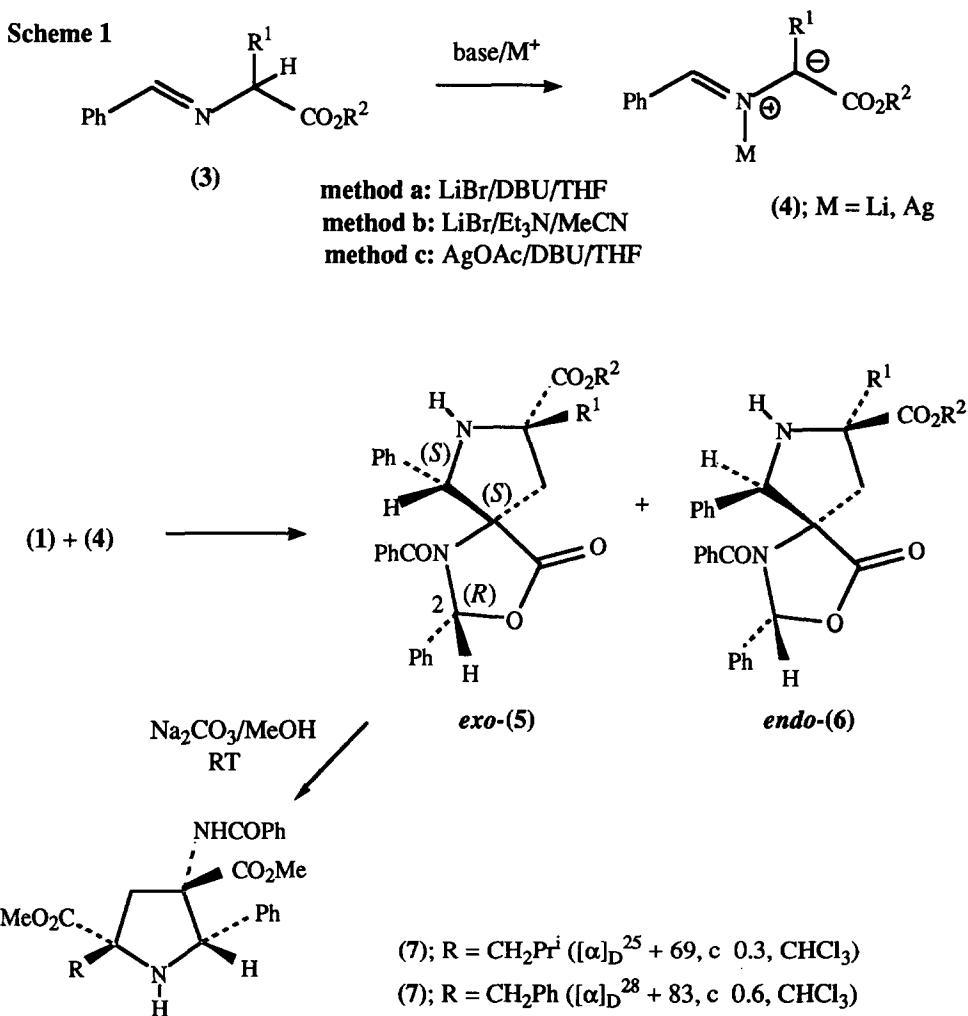
Abstract: *Exo*-diastereoselective 1,3-dipolar cycloadditions of azomethine ylides derived from α -amino acid esters to (2*R*)-3-Benzoyl-4-methylene-2-phenyloxazolidin-5-one (1) are reported. The cycloaddition products are conveniently converted to polyfunctional prolines in high enantiomeric purity.

The 1,3-dipolar cycloaddition reactions of azomethine ylides derived from *N*-alkylidene α -amino acid esters to electron deficient alkenes is an extremely powerful method for the synthesis of polyfunctional prolines.¹⁻⁵ The asymmetric version of these reactions has been realised by employing a chiral auxiliary attached to either the dipolarophile³ or the azomethine ylide⁴ or the use of a chiral metal complex catalyst.⁵ As part of a continuing program we have been interested in the asymmetric synthesis of novel non-proteinogenic amino acids using the chiral oxazolidinones (1)⁶ and (2).⁶⁻⁸ We have recently reported that (1) and (2) undergo highly *exo*-diastereoselective Diels-Alder reaction with dienes to give chiral carbocyclic amino acid derivatives in high enantiomeric purities.^{6,9} Other workers have demonstrated that (2) undergoes diastereoselective 1,2-addition reactions with alkyl radicals¹⁰ and nitronate anions¹¹ and cyclopropanation reactions.¹² We now report that (1) undergoes highly regioselective and *exo*-diastereoselective 1,3-dipolar cycloaddition reactions with azomethine ylides derived from *N*-benzylidene α -amino acid esters. The cycloaddition products are conveniently converted to polyfunctional prolines in high enantiomeric purity.



It is well documented that azomethine ylides can be prepared *in situ* from the reaction of *N*-alkylidene α -amino acid esters and a variety of metal cations in the presence of a base.^{2,3(a-f)} In this study with (1) and imines (3), we found that the metal salt/base/solvent combination of LiBr/DBU/THF^{2(a,b,e,g),3(a-e)} was far superior to AgOAc/DBU/THF^{2(b,d-g),3(b,f)} or LiBr/Et₃N/MeCN^{2(a,b,e,g),3(a-e)} in terms of chemical reactivity and product diastereoselectivity. The results of this study are summarised in Table 1. The azomethine ylides

(4) were generated *in situ* from (3) in the presence of (1) by treating a THF or MeCN solution of (1), (3) and the metal salt with base (DBU or Et₃N) at -78°C. The reaction mixtures were maintained at -78°C for several hours or warmed to the temperatures specified in Table 1 before being quenched with saturated aqueous NH₄Cl solution. In each case the diastereoselectivities of these reactions were measured on the crude reaction products by ¹H NMR analysis (Typically H₂ in (5) (δ 5.76–5.88, s) was observed upfield of H₂ in (6) (δ 6.04–6.35, s)). In each case a mixture of *exo*-(5) and *endo*-(6) products were formed and these could be conveniently separated by column chromatography. The yields in Table 1 refer to the combined yield of (5) and (6) after chromatographic separation.



In all but one case (Table 1, entry 6) the reactions were completely regioselective and showed good to high diastereoselectivity in favour of the *exo*-diastereoisomer (5) over the *endo*-diastereoisomer (6). The structures of (5) (R¹ = CH₂Prⁱ) and (6) (R¹ = CH₂Prⁱ and CH₂Ph) were determined by single-crystal X-ray structural analysis and the structural assignments of the other adducts (5) and (6) (R¹ = H, Me, Ph and CH₂Ph) were

based on the similarity of their ^1H NMR spectra to those of (5) ($\text{R}^1 = \text{CH}_2\text{Pr}^i$) and (6) ($\text{R}^1 = \text{CH}_2\text{Pr}^i$). When $\text{LiBr}/\text{Et}_3\text{N}/\text{MeCN}$ was used the diastereoselectivities were generally lower except in the case of (3) ($\text{R}^1 = \text{Ph}$) (Table 1, entry 8) where an enhanced *exo*-diastereoselectivity (93 : 7) was realized. When AgOAc/DBU in THF was employed the diastereoselectivity was in general much lower (64-55 : 36-45) and the reactions did not proceed at an appreciable rate below 0°C . In the case of the glycine derivative (3) ($\text{R}^1 = \text{H}$) (Table 1, entry 12), the $\text{LiBr}/\text{Et}_3\text{N}$ or DBU method gave two as yet unknown structures, while a 43 : 57 mixture of (5) ($\text{R}^1 = \text{H}$) and (6) ($\text{R}^1 = \text{H}$) resulted when AgOAc/DBU in THF was used.

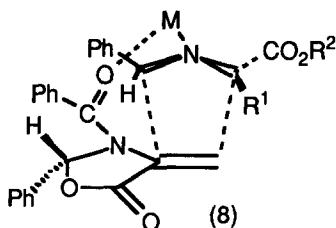
The adducts (5) ($\text{R}^1 = \text{CH}_2\text{Pr}^i$, CH_2Ph) were converted to the highly functionalised proline derivatives (7) ($\text{R} = \text{CH}_2\text{Pr}^i$, CH_2Ph) upon mild treatment with sodium carbonate in methanol. The enantiomeric purity of (7) ($\text{R} = \text{CH}_2\text{Pr}^i$) was determined to be 92% based on ^1H NMR analysis of its diastereomeric carbamate derivatives from treatment with (S)-(+)-1-phenylethylisocyanate.

Table 1. 1,3-Dipolar cycloaddition products from azomethine ylides (4) and (1).

Entry	Imine (2)		Method	Time(h)/Temp($^\circ\text{C}$)	Yield(%)	Diastereoselectivity (5) : (6)
	R^1	R^2				
1	Me	Me	a	6/-78	52	82 : 18
2	Me	Me	b	8/0	69	76 : 24
3	Me	Me	c	20/RT	83	58 : 42
4	CH_2Pr^i	Me	a	30/-20	61	94 : 6
5	CH_2Pr^i	Me	b	10/0	62	82 : 18
6	CH_2Pr^i	Me	c	30/RT	67	57 : 37 : 6
7	Ph	Me	a	20/-78	50	83 : 17
8	Ph	Me	b	20/0	59	93 : 7
9	CH_2Ph	Et	a	9/-78	45	85 : 15
10	CH_2Ph	Et	b	3/0	51	78 : 22
11	CH_2Ph	Et	c	30/RT	51	64 : 36
12	H	Et	c	6/RT	88	43 : 57

In summary, we report that the 1,3-dipolar cycloaddition reactions of (1) and the azomethine ylides (4) proceed with good to high *exo*-diastereoselectivity and that the cycloaddition adducts can be converted to highly functionalised prolines in high enantiomeric purity. The preference for *exo*-cycloaddition adducts can be rationalised by assuming chelation between the lithium cation and the *N*-benzoyl carbonyl group and the azomethine ylide (4), as shown in the possible transition state structure (8). While *endo*-cycloaddition adducts

are generally favoured in these types of reactions, 1,2(a,e,g),3(a,c-e,g),4(b,c,g,h) *exo*-diastereoselective azomethine ylide cycloadditions are not uncommon.^{4(e),13}



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